What happens to the melting point as you move down group 1?	Describe the trend in reactivity of the group 7 elements.	Give the <b>names</b> of Groups 1, 7, 0 and cold Identify the metals and non-metals.	our them in different colours on the table below.
Define a displacement reaction	Give an example of an element that could be used in electrical wiring (the element must be able to conduct electricity).	Periodic Tab	le of the Elements
Describe the trend in reactivity of the group elements	Write a word equation for the reaction of o	chlorine with potassium bromide.	What element would we use to fill a hot air balloon? And why?
Write a word equation for the reaction of sodium with water. Describe the observations you would see.	Write a word equation for the reaction of Chlorine with Iron.  + →		Explain why bromine can displace iodine from lithium iodide.
	Describe the trend in melting & boiling points of group 7.	What is the difference between the reactivity trend in Group 7 and Group 1?	
How would the observations differ if you added: Lithium?  Potassium?			How is the modern periodic table arranged?